



Cambridge International AS & A Level

PHYSICS

9702/21

Paper 2 AS Level Structured Questions

October/November 2022

MARK SCHEME

Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2022 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

This document consists of **12** printed pages.

PUBLISHED**Generic Marking Principles**

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
- 5 'List rule' guidance

For questions that require ***n*** responses (e.g. State **two** reasons ...):
 - The response should be read as continuous prose, even when numbered answer spaces are provided.
 - Any response marked *ignore* in the mark scheme should not count towards ***n***.
 - Incorrect responses should not be awarded credit but will still count towards ***n***.
 - Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
 - Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Abbreviations

/	Alternative and acceptable answers for the same marking point.
()	Bracketed content indicates words which do not need to be explicitly seen to gain credit but which indicate the context for an answer. The context does not need to be seen but if a context is given that is incorrect then the mark should not be awarded.
—	Underlined content must be present in answer to award the mark. This means either the exact word or another word that has the same technical meaning.

Mark categories

B marks	These are <u>independent</u> marks, which do not depend on other marks. For a B mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer.
M marks	These are <u>method</u> marks upon which A marks later depend. For an M mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer. If a candidate is not awarded an M mark, then the later A mark cannot be awarded either.
C marks	These are <u>compensatory</u> marks which can be awarded even if the points to which they refer are not written down by the candidate, providing subsequent working gives evidence that they must have known them. For example, if an equation carries a C mark and the candidate does not write down the actual equation but does correct working which shows the candidate knew the equation, then the C mark is awarded. If a correct answer is given to a numerical question, all of the preceding C marks are awarded automatically. It is only necessary to consider each of the C marks in turn when the numerical answer is not correct.
A marks	These are <u>answer</u> marks. They may depend on an M mark or allow a C mark to be awarded by implication.

Question	Answer	Marks
1(a)	any two joined correctly	C1
	all four joined correctly	A1
1(b)(i)	the measurements have a small range	B1
1(b)(ii)	(average of the) measurements not close to the true value	B1
1(c)(i)	percentage uncertainty = $(3 + 5 + 4) / 2$	C1
	= 6%	A1
1(c)(ii)	absolute uncertainty = $(6 / 100) \times 15.0$ = 0.9 m s^{-1}	A1

PUBLISHED

Question	Answer	Marks
2(a)(i)	$t = 1.8 / 4.9$ $= 0.37 \text{ s}$	A1
2(a)(ii)	$v = u + at$ $= 9.81 \times 0.37$	C1
	$= 3.6 \text{ m s}^{-1}$	A1
2(a)(iii)	$v^2 = 3.6^2 + 4.9^2$	C1
	$v = 6.1 \text{ m s}^{-1}$	A1
	$\theta = \tan^{-1} (3.6 / 4.9)$ $= 36^\circ$	A1
2(b)(i)	$\rho = m / V$	C1
	$V = \frac{4}{3} \pi r^3$	C1
	$\rho = 0.017 / \left[\frac{4}{3} \pi \times (0.016 / 2)^3 \right]$ $= 7900 \text{ kg m}^{-3}$	A1
2(b)(ii)	$(E =) \frac{1}{2}mv^2$	C1
	$(E =) \frac{1}{2} \times 0.017 \times 4.9^2 = 0.20 \text{ (J)}$	A1

PUBLISHED

Question	Answer	Marks
2(b)(iii)	$k = F/x$ or $k = \text{gradient}$	C1
	e.g. $k = 6.4 / 10 \times 10^{-2}$ $= 64 \text{ N m}^{-1}$ (allow 63–65 N m^{-1})	A1
2(b)(iv)	$E = \frac{1}{2}kx^2$ or $E = \frac{1}{2}Fx$ and $F = kx$	C1
	$x_0 = [(2 \times 0.20) / 64]^{0.5}$ $= 0.079 \text{ m or } 0.080 \text{ m}$	A1
2(c)(i)	same elastic potential energy / same (initial) kinetic energy and (polystyrene ball has) smaller mass (so greater speed) or same (average) force and (polystyrene ball has) smaller mass, (so greater average acceleration so greater speed)	B1
2(c)(ii)	(for the polystyrene ball there is) less (average vertical) acceleration / smaller (average vertical component of) <u>resultant</u> force (so takes longer time to reach ground)	B1

PUBLISHED

Question	Answer	Marks
3(a)(i)	work done per unit time	B1
3(a)(ii)	$W = Fs$	B1
	$P = Fs / t$ <u>and</u> (so) $P = Fv$	B1
3(b)	$(F \Rightarrow) 130 \times 10^3 / 25 = 5200 \text{ (N)}$	A1
3(c)(i)	(component of weight $\Rightarrow$) $mg \sin \theta$	A1
3(c)(ii)	$F \text{ (along slope due to weight)} = 36\,000 \times 9.81 \times \sin 1.4^\circ$ $(= 8600 \text{ N})$	C1
	$(\text{total}) F = 5200 + 36\,000 \times 9.81 \times \sin 1.4^\circ$ $(= 13\,800 \text{ N})$	C1
	$P = 13\,800 \times 25$ $= 350 \times 10^3 \text{ (W)}$ $= 350 \text{ kW}$	A1

PUBLISHED

Question	Answer	Marks
4(a)(i)	oscillations are in a single direction, which is perpendicular to the direction of propagation (of the wave) or oscillations are in a single plane, which contains the direction of propagation (of the wave)	B1
4(a)(ii)	light waves are transverse and sound waves are longitudinal	B1
4(b)	$I = I_0 \cos^2 \theta$	C1
	$\cos^2 \theta = 1/4$ so $\cos \theta = 1/2$ $\theta = 60^\circ$ or 120° or 240° or 300°	C1
	angle of rotation = $(120^\circ - 90^\circ)$ or $(240^\circ - 90^\circ)$ or $(300^\circ - 90^\circ)$ $= 30^\circ$ or 150° or 210° or 330°	A1
4(c)(i)	the waves have different amplitudes	B1
	cannot have resultant displacement that is always zero or cannot have (complete) destructive interference (at nodes) or (at nodes resultant) amplitude is the difference of the amplitudes	B1
4(c)(ii)	$I \propto A^2$	C1
	$A^2 / A_0^2 = (I_0 / 4) / I_0$ $A = 0.5 A_0$	C1
	maximum amplitude = $A_0 + 0.5 A_0$ $= 1.5 A_0$	A1

PUBLISHED

Question	Answer	Marks
5(a)	current (through a conductor is directly) proportional to potential difference (across the conductor)	M1
	(provided that) temperature (of conductor remains) constant	A1
5(b)(i)	(ratio of) V/I increases (as p.d. increases)	B1
5(b)(ii)	(as p.d. increases, current increases so) temperature increases	B1
5(c)(i)	$I = 1.55 \text{ A}$	A1
5(c)(ii)	$P = VI$ or $P = I^2R$ or $P = V^2/R$	C1
	$= 6.0 \times 1.55 \times 2$ or $1.55^2 \times 3.87 \times 2$ or $(6.0^2/3.87) \times 2$ $= 19 \text{ W}$	A1
5(c)(iii)	$I = 1.78 - 1.55$ (= 0.23 A)	C1
	$R = 12.0 / 0.23$ $= 52 \Omega$	A1
5(d)	lamp P: p.d. across lamp decreases to zero so goes 'out'	B1
	lamp Q: p.d. across lamp increases to 12 V so gets brighter	B1

PUBLISHED

Question	Answer	Marks
6(a)	particle with no internal structure / particle which cannot be broken down into anything smaller	A1
6(b)	charges: $u = (+)\frac{2}{3}(e)$ or $d = -\frac{1}{3}(e)$ or $s = -\frac{1}{3}(e)$	C1
	$(+)\frac{2}{3}(e) - \frac{1}{3}(e) - \frac{1}{3}(e) = 0(e)$	A1
6(c)(i)	• same/equal mass • same/equal (magnitude of) charge • both fundamental (particles) • opposite (sign of) charge • one is matter and the other is antimatter <i>Any two points, 1 mark each.</i>	B2
6(c)(ii)	neutron/baryon consists of three quarks	B1
	pion/meson consists of one quark and one antiquark	B1



Cambridge International AS & A Level

PHYSICS

9702/22

Paper 2 AS Level Structured Questions

October/November 2022

MARK SCHEME

Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2022 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

This document consists of **12** printed pages.

PUBLISHED**Generic Marking Principles**

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
- 5 'List rule' guidance

For questions that require ***n*** responses (e.g. State **two** reasons ...):
 - The response should be read as continuous prose, even when numbered answer spaces are provided.
 - Any response marked *ignore* in the mark scheme should not count towards ***n***.
 - Incorrect responses should not be awarded credit but will still count towards ***n***.
 - Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
 - Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Abbreviations

/	Alternative and acceptable answers for the same marking point.
()	Bracketed content indicates words which do not need to be explicitly seen to gain credit but which indicate the context for an answer. The context does not need to be seen but if a context is given that is incorrect then the mark should not be awarded.
<u> </u>	Underlined content must be present in answer to award the mark. This means either the exact word or another word that has the same technical meaning.

Mark categories

B marks	These are <u>independent</u> marks, which do not depend on other marks. For a B mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer.
M marks	These are <u>method</u> marks upon which A marks later depend. For an M mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer. If a candidate is not awarded an M mark, then the later A mark cannot be awarded either.
C marks	These are <u>compensatory</u> marks which can be awarded even if the points to which they refer are not written down by the candidate, providing subsequent working gives evidence that they must have known them. For example, if an equation carries a C mark and the candidate does not write down the actual equation but does correct working which shows the candidate knew the equation, then the C mark is awarded. If a correct answer is given to a numerical question, all of the preceding C marks are awarded automatically. It is only necessary to consider each of the C marks in turn when the numerical answer is not correct.
A marks	These are <u>answer</u> marks. They may depend on an M mark or allow a C mark to be awarded by implication.

PUBLISHED

Question	Answer	Marks
1(a)	force $\times$ displacement in the direction of the force	B1
1(b)	$P = Fs / t$ $= (\text{kg m s}^{-2} \times \text{m}) / \text{s}$	C1
	$= \text{kg m}^2 \text{s}^{-3}$	A1
1(c)(i)	$84 \times 10^3 = v^3 \times 0.56$	C1
	$v = 53 \text{ m s}^{-1}$	A1
1(c)(ii)	percentage uncertainty = $(5\% + 7\%) / 3$ (= 4%) or fractional uncertainty = $(0.05 + 0.07) / 3$ (= 0.04)	C1
	absolute uncertainty = 0.04×53 $= (\pm) 2 \text{ m s}^{-1}$	A1

Question	Answer	Marks
2(a)	$F = \rho g V$ $0.071 = 1.2 \times 9.81 \times V$	C1
	$(V = 6.03 \times 10^{-3} \text{ m}^3)$ $\frac{4}{3} \times \pi \times r^3 = 6.03 \times 10^{-3}$ $r = 0.11 \text{ m}$	A1
2(b)	$m = 0.053 / 9.81$ $(= 5.4 \times 10^{-3} \text{ kg})$	C1
	$F = 0.071 - 0.053$ $(= 0.018 \text{ N})$	C1
	$a = (0.071 - 0.053) / (0.053 / 9.81)$ $= 3.3 \text{ m s}^{-2}$	A1
2(c)(i)	$v^2 = u^2 + 2as$ $3.6^2 = (-1.4)^2 + 2 \times 9.81 \times s$ or $3.6^2 = 1.4^2 + 2 \times 9.81 \times s$	C1
	$s = 0.56 \text{ m}$	A1
2(c)(ii)	single straight line from any positive non-zero value of v at $t = 0$ to any negative non-zero value of v at $t = T$	B1
	line starting at $(0, 1.4)$ and ending at $(T, -3.6)$	B1

PUBLISHED

Question	Answer	Marks
3(a)	point where (all) the weight (of an object) is taken to act	B1
3(b)(i)	vertical component = $45 \sin 37^\circ$ = 27 N	A1
3(b)(ii)	the magnitudes of the three moments about A are (23×0.48) , (27×0.56) and $(W \times 0.76)$	C1
	correct magnitude of any one moment about A	
	correct magnitudes of any two moments about A	C1
	$(23 \times 0.48) + (W \times 0.76) = 27 \times 0.56$ $W = 5.4 \text{ N}$	A1
3(b)(iii)	horizontal component = $45 \cos 37^\circ$ = 36 N	A1
3(b)(iv)	decrease	B1
3(b)(v)	$\sigma = F / A$	C1
	$\sigma = F / \pi r^2$ or $4F / \pi d^2$ so $\sigma = 5.3 \times 10^7 \times \pi r^2 / \pi (3r)^2$ = $5.3 \times 10^7 / 9$ = $5.9 \times 10^6 \text{ Pa}$	A1

PUBLISHED

Question	Answer	Marks
4(a)	$E = \frac{1}{2}mv^2$	C1
	$= \frac{1}{2} \times 0.25 \times 2.3^2$	A1
	$= 0.66 \text{ J}$	
4(b)	$E = \frac{1}{2}kx^2$ or $E = \frac{1}{2}Fx$ and $F = kx$	C1
	$x = [(2 \times 0.66) / 420]^{0.5}$ $= 0.056 \text{ m}$	A1
	or	
	$E = \frac{1}{2}kx^2$	(C1)
	$\frac{1}{2}mv^2 = \frac{1}{2}kx^2$ $x = (0.25 \times 2.3^2 / 420)^{0.5}$ $= 0.056 \text{ m}$	(A1)
4(c)(i)	$(p =) 0.25 \times 2.3$ or 0.25×1.5	C1
	change in momentum $= 0.25 (2.3 + 1.5)$ $= 0.95 \text{ N s}$	A1
4(c)(ii)	resultant force $= 0.95 / 0.086$ or $0.25 \times (2.3 + 1.5) / 0.086$ $= 11 \text{ N}$	A1
4(d)	curved line from the origin with an increasing gradient	B1

PUBLISHED

Question	Answer	Marks
5(a)	(they travel in) opposite directions	B1
5(b)(i)	straight line from A to B, labelled P	B1
	line that is 'mirror image' of given line, labelled Q	B1
5(b)(ii)	$\lambda = 0.80 / 2$ (= 0.40 m)	C1
	$v = \lambda / T$ or $v = f\lambda$ and $f = 1 / T$	C1
	$v = 0.40 / 0.016$ or 62.5×0.40 $= 25 \text{ m s}^{-1}$	A1
5(c)(i)	$I_1 / I_0 = \cos^2 30^\circ$	C1
	$= 0.75$	A1
5(c)(ii)	$I_2 / I_1 = \cos^2 60^\circ$ $I_2 / I_0 = \cos^2 30^\circ \times \cos^2 60^\circ$ or $0.75 \times \cos^2 60^\circ$	C1
	$= 0.19$	A1

PUBLISHED

Question	Answer	Marks
6(a)	energy transferred per (unit) charge (from electrical to other forms)	B1
6(b)	same/equal current (in X and Y)	B1
	$P = I^2 R$ (and $R_X > R_Y$) or $P = VI$ and $V_X > V_Y$	M1
	(so) X (dissipates more power)	A1
6(c)(i)	$I = Q / t$	C1
	$= 650 / 540$	A1
	$= 1.2 \text{ A}$	
6(c)(ii)	$V = W / Q$ or W / It	C1
	$= 4800 / 650$ or $4800 / (1.2 \times 540)$	A1
	$= 7.4 \text{ V}$	
	or	
	$V = P / I$ and $P = W / t$	(C1)
	$= 8.9 / 1.2$	(A1)
	$= 7.4 \text{ V}$	
6(c)(iii)	$I = 4.5 + 1.2 (= 5.7 \text{ A})$	C1
	$9.0 = 7.4 + 5.7r$ or $9.0 = 5.7 (1.3 + r)$	A1
	$r = 0.28 \Omega$	

PUBLISHED

Question	Answer	Marks
7(a)	92 protons and 146 neutrons (in nucleus)	B1
	92 (orbital) electrons	B1
7(b)	charge = $2e$ (= $2 \times 1.60 \times 10^{-19} \text{ C}$)	C1
	mass = $4u$ (= $4 \times 1.66 \times 10^{-27} \text{ kg}$)	C1
	ratio = $(2 \times 1.60 \times 10^{-19}) / (4 \times 1.66 \times 10^{-27})$ = $4.8 \times 10^7 \text{ C kg}^{-1}$	A1
7(c)(i)	up down down / udd	B1
7(c)(ii)	up up up / uuu	B1



Cambridge International AS & A Level

PHYSICS

9702/23

Paper 2 AS Level Structured Questions

October/November 2022

MARK SCHEME

Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2022 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

This document consists of **14** printed pages.

PUBLISHED**Generic Marking Principles**

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
- 5 'List rule' guidance

For questions that require ***n*** responses (e.g. State **two** reasons ...):
 - The response should be read as continuous prose, even when numbered answer spaces are provided.
 - Any response marked *ignore* in the mark scheme should not count towards ***n***.
 - Incorrect responses should not be awarded credit but will still count towards ***n***.
 - Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
 - Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Abbreviations

/	Alternative and acceptable answers for the same marking point.
()	Bracketed content indicates words which do not need to be explicitly seen to gain credit but which indicate the context for an answer. The context does not need to be seen but if a context is given that is incorrect then the mark should not be awarded.
—	Underlined content must be present in answer to award the mark. This means either the exact word or another word that has the same technical meaning.

Mark categories

B marks	These are <u>independent</u> marks, which do not depend on other marks. For a B mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer.
M marks	These are <u>method</u> marks upon which A marks later depend. For an M mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer. If a candidate is not awarded an M mark, then the later A mark cannot be awarded either.
C marks	These are <u>compensatory</u> marks which can be awarded even if the points to which they refer are not written down by the candidate, providing subsequent working gives evidence that they must have known them. For example, if an equation carries a C mark and the candidate does not write down the actual equation but does correct working which shows the candidate knew the equation, then the C mark is awarded. If a correct answer is given to a numerical question, all of the preceding C marks are awarded automatically. It is only necessary to consider each of the C marks in turn when the numerical answer is not correct.
A marks	These are <u>answer</u> marks. They may depend on an M mark or allow a C mark to be awarded by implication.

Question	Answer	Marks
1(a)	(SI base unit of α =) $\text{m}^3 \text{s}^{-1} \times \text{m} / \text{m}^4 = \text{s}^{-1}$	A1
1(b)	(percentage uncertainty =) $3 + 4 + 2 \times 4 = 15$ (%)	A1
1(c)	$\alpha = QL / r^4$ $= 2.72 \times 10^{-8} \times 2.5 \times 10^{-2} / (7.1 \times 10^{-5})^4$ $= 2.7 \times 10^7$	C1
	absolute uncertainty = $0.15 \times [2.7 \times 10^7]$ $= 0.4 \times 10^7$	C1
	$\alpha = (2.7 \pm 0.4) \times 10^7 \text{ s}^{-1}$	A1

PUBLISHED

Question	Answer	Marks
2(a)(i)	(e.g. $a =$) $[16.5 - 0] / [0.30 - 0] = 55 \text{ (m s}^{-2}\text{)}$	A1
2(a)(ii)	$(F =) T - W$	A1
2(a)(iii)	$ma = T - mg$ $m = 16 / (55 + 9.81)$	C1
	$m = 0.25 \text{ kg}$	A1
2(b)(i)	$s = ut + \frac{1}{2}at^2$ or $v^2 = u^2 + 2as$ or $s = vt - \frac{1}{2}at^2$ or $s = \frac{1}{2}(u + v)t$ or $s = \text{area under graph}$ $s = \frac{1}{2} \times 55 \times 0.30^2$ or $s = 16.5^2 / (2 \times 55)$ or $s = 16.5 \times 0.30 - \frac{1}{2} \times 55 \times 0.30^2$ or $s = \frac{1}{2} \times 16.5 \times 0.30$	C1
	$s = 2.5 \text{ m}$	A1
2(b)(ii)	$u = 16.5 \text{ (m s}^{-1}\text{)}$	C1
	$v^2 = u^2 + 2as$ $v^2 = 16.5^2 + 2 \times 9.81 \times 2.5$	C1
	$v = 18 \text{ m s}^{-1}$	A1

PUBLISHED

Question	Answer	Marks
3(a)	sum of CW moments = sum of ACW moments	M1
	about the same point for (an object in rotational) equilibrium	A1
3(b)	moment = $0.3(0) \times 0.29 \cos 40^\circ$ or $0.3(0) \times 0.222$	C1
	= 0.067 N m	A1
3(c)(i)	$k = F/x$ or $k = \text{gradient}$	C1
	e.g. $k = 21/10 \times 10^{-3}$ $k = 2100 \text{ N m}^{-1}$	A1
3(c)(ii)	$V_{\text{sphere}} = \frac{4}{3} \times \pi \times (0.0480)^3$	C1
	$F = \rho g V$ (upthrust =) $1000 \times 9.81 \times \left(\frac{4}{3} \times \pi \times (0.048)^3\right) \times 0.26(0) = 1.18 \text{ (N)}$	A1
3(c)(iii)	1.18×0.29 or 0.30×0.29 or $F \times 0.017$	C1
	$(1.18 \times 0.29) = (0.30 \times 0.29) + (F \times 0.017)$ $F = 15 \text{ N}$	A1

PUBLISHED

Question	Answer	Marks
3(c)(iv)	$E_{(P)} = \frac{1}{2}kx^2$ or $E_{(P)} = \frac{1}{2}Fx$	C1
	$x = F/k = 15/2100$ or x determined from graph for $F = 15.0$ N	A1
	$E_P = \frac{1}{2} \times 2100 \times (15/2100)^2$ or $E_P = \frac{1}{2} \times 15 \times (15/2100)$	
	$E_P = 0.054$ J	
3(d)	the <u>sphere</u> has gained gravitational potential energy	B1

PUBLISHED

Question	Answer	Marks
4(a)(i)	circles drawn around any two particles with seven (uncircled) particles in between	A1
4(a)(ii)	curve has an initial negative displacement and initial amplitude same as original curve	B1
	curve has same amplitude as original curve throughout	B1
	curve has same wavelength as original curve throughout, with constant (non-zero) phase difference	B1
4(a)(iii)	(to the) right / rightwards	A1
4(b)	$\lambda = 2 \times 0.19$ $= 0.38 \text{ m}$	C1
	$v = f\lambda$	C1
	$v = (16 \times 10^3) \times 0.38$ $v = 6100 \text{ m s}^{-1}$	A1
4(c)	$I \propto A^2$	C1
	$I = (1/2)^2 I_0$ intensity = $0.25 I_0$	A1
4(d)(i)	same frequency / wavelength / period	B1
4(d)(ii)	- a stationary wave has nodes/antinodes (and a progressive wave does not) - a stationary wave does not transfer/propagate energy (and a progressive wave does transfer/propagate energy) - different points on a stationary wave have different amplitudes (and all points on a progressive wave have the same/constant amplitude) - stationary wave has adjacent particles that are in phase (and adjacent particles on progressive wave are out of phase) <i>Any two points, 1 mark each. Allow the reverse statement for each marking point.</i>	B2

PUBLISHED

Question	Answer	Marks
5(a)	sum of e.m.f.(s) = sum of p.d.(s) or (algebraic) sum of e.m.f.(s) and p.d.(s) is zero	M1
	around a loop / around a <u>closed</u> circuit	A1
5(b)(i)	$1 / r_{(T)} = 1 / 0.59 + 1 / 0.59 + 1 / 0.59$	B1
	$(r_{(T)} =) 0.197 \, (\Omega)$ $(R =) 2.2 - 0.197 = 2.0 \, \Omega$	A1
	or	
	$I = 1.5 / 2.2 (= 0.68 \, \text{A})$ and $i = 0.68 / 3$ (where I is the circuit current and i is the current from each cell)	(B1)
	$(E = IR + ir =) 1.5 = 0.68R + (0.68 / 3) \times 0.59$ and $R = 2.0 \, \Omega$	(A1)

PUBLISHED

Question	Answer	Marks
5(b)(ii)	current = $1.5 / 2.2$	C1
	= 0.68 A	A1
	or	
	p.d. across cell = p.d. across conductor $1.5 - 0.59I = 3I \times 2.0$ so $I = 0.228$ A (where I is current in cell) current = 3×0.228	(C1)
	= 0.68 A	(A1)
	or	
	current in conductor = $3 \times$ current in cell $V / 2.0 = 3 \times (1.5 - V) / 0.59$ (where V is p.d. across conductor) $V = 1.37$ V current = $1.37 / 2.0$	(C1)
	= 0.68 A	(A1)
5(c)(i)	$R = \rho L / A$	C1
	$R = 4\rho L / \pi d^2$ (ρ and L are the same so) $R_A / R_B = 7.6^2 / 4.3^2$	C1
	= 3.1	A1

PUBLISHED

Question	Answer	Marks
5(c)(ii)	$I = Anvq$ and I, n, q are same / equal / constant	B1
	$\frac{v_A}{v_B} = \frac{A_B}{A_A} = \frac{d_B^2}{d_A^2}$ ratio = $7.6^2 / 4.3^2$ = 3.1	A1
5(d)	combined internal resistance (of the cells) will be greater or total / circuit resistance (of circuit) greater (because a parallel resistance removed)	B1
	more 'lost volts' (inside each cell) or internal resistances take a greater share of total p.d. or conductor gets a smaller share of the total p.d. or current in <u>conductor</u> /total current decreases	M1
	(so) potential difference (across conductor) decreases	A1

PUBLISHED

Question	Answer	Marks
6(a)	${}^{14}_7\text{X}$	B1
	${}^0_{-1}\text{e}^-$	B1
6(b)(i)	$\text{d} \rightarrow \text{u} + \text{e}^- + \bar{\nu}$ or $\text{udd} \rightarrow \text{uud} + \text{e}^- + \bar{\nu}$	B1
6(b)(ii)	$-1/3 (e) = +2/3 (e) - 1(e) (+0)$ or $2/3 (e) - 1/3 (e) - 1/3 (e) = 2/3 (e) + 2/3 (e) - 1/3 (e) - 1(e) (+0)$	B1
6(c)(i)	electrons / β -particles (emitted from the nucleus) have a (continuous) range of / different (kinetic) energies	B1
6(c)(ii)	the (emitted) neutrinos take varying amounts of the (same total) energy (released in the decay)	B1